

common type non-ionics while on cotton the efficiency seems to be poorer at the lower concentrations but at use concentrations again to be comparable with other type non-ionics.

The sudsing characteristics of the wax acid non-ionics were observed in launderometer tests as well as in many of the empirical shaking tests. In most instances the amount of suds formed was small and not too lasting.

Testing of Drying Oils. II. Evaluation of Natural and Synthetic Oils*

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IN a previous paper Bolley and Gallagher (1) have described in detail a method for the preliminary examination of drying oils. This method has been used by the National Lead Company's Research Laboratory for the evaluation of a large number of oils. It is the purpose of this paper to present some of the data obtained during the evaluation program.

Since this is a preliminary evaluation which is to be completed in a relatively short time, the scheme does not include exposure tests. The object is to characterize the oils to determine their probable application; final evaluation must await exposure tests. Accelerated exposure tests were not included since these have been found to be unreliable in many cases.

One test has been added to the evaluation program as previously described. This is the determination of flash point according to the standard A.S.T.M. Method D92-33 and was included when it was noticed that some synthetic oils had relatively low flash points. A comparison of the data presented in this paper with the previously published method will disclose that not all of the tests have been reported. Although all the tests were actually run on the oils, it seemed desirable to eliminate the results of lesser interest and significance in order to keep the already lengthy tables of data to a minimum.

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Conclusion

Non-ionic synthetic detergents having detergent properties similar to those of alkyl phenol polyethylene glycol ethers and alkyl mercapto polyethylene glycol ethers can be prepared by reacting the optimum quantity of ethylene oxide with the composite crude fatty acids made by the air oxidation of low melting chemical grade paraffin wax.

The following oils, referred to in the abbreviated form used in the tables, were tested:

G Bodied Linseed—An alkali refined, bleached, and refrigerated linseed oil heat bodied to a G viscosity.

Conjugated Linseed—A refined linseed oil catalytically treated to contain 8.1% conjugated linoleic isomer and 0.5% conjugated linolenic isomer.

Linseed Penta.—Technical pentaerythritol esterified with nearly equivalent amounts of linseed oil fatty acids.

Dehydrated Castor—Regular commercial unbodied dehydrated castor oil.

Fatty Tall Oil—Tall oil processed to remove a large amount of the rosin acids and the fatty acid concentrate esterified with technical pentaerythritol.

G Bodied Soybean—Refined soybean oil heat bodied to a G viscosity.

Conjugated Soybean—A refined soybean oil catalytically treated to contain 10.1% conjugated linoleic isomer and 1.2% conjugated linolenic isomer.

Soybean Penta.—Technical pentaerythritol esterified with nearly equivalent amounts of soybean oil fatty acids.

Soybean Mannitol—One mol. of mannitol esterified with four mols. of soybean oil fatty acids.

Soybean Sorbitol—One mol. of sorbitol esterified with four mols. of soybean oil fatty acids.

Q Bodied Linseed—An alkali refined, bleached, and refrigerated linseed oil heat bodied to a Q viscosity.

Linseed Polyenta.—Technical polyentaerythritol esterified with nearly equivalent amounts of linseed oil fatty acids.

Linseed Mannitol—One mol. of mannitol esterified with four mols. of linseed oil fatty acids.

Linseed Sorbitol—One mol. of sorbitol esterified with four mols. of linseed oil fatty acids.

TABLE I
Chemical Constants

	Visc.	Color	Appearance	Odor	Acid Value	Sap. Value	Acetyl Value	Iodine Value	% Unsap.	% Ash	Ref. Index	Specific Gravity
G Bodied Linseed.....	G	6	Clear	Normal	2.1	189.0	5.4	165.0	1.43	0.000	1.4841	0.9430
Conjugated Linseed.....	F	7	Clear	Bodied Oil	3.3	190.7	7.4	154.7	1.20	0.000	1.4838	0.9431
Linseed Penta.....	I-	12-	Clear	Bodied Oil	4.0	180.9	28.4	155.6	1.99	0.000	1.4850	0.9524
Dehydrated Castor.....	H+	6	Clear	Normal	8.5	198.3	1.6	137.3	0.96	0.000	1.4820	0.9377
Fatty Tall Oil.....	J-	10	Clear	Fatty	6.7	170.8	3.6	133.1	4.80	0.055	1.4861	0.9490
G Bodied Soybean.....	F	5+	Clear	Normal	2.2	189.6	5.0	119.0	1.15	0.000	1.4765	0.9358
Conjugated Soybean.....	G	6	Clear	Normal	3.2	192.3	3.4	118.5	0.92	0.000	1.4771	0.9356
Soybean Penta.....	II+	12	Clear	Normal	3.3	185.9	52.2	125.1	1.00	0.000	1.4796	0.9475
Soybean Mannitol.....	G	7-	Clear	Normal	29.5	183.2	3.1	112.1	0.71	0.000	1.4774	0.9402
Soybean Sorbitol.....	F	6	Clear	Normal	16.5	185.1	7.8	119.6	0.80	0.020	1.4775	0.9463
Q Bodied Linseed.....	U	5	Clear	Bodied Oil	5.9	192.2	6.9	141.4	1.06	0.000	1.4858	0.9585
Linseed Polyenta.....	Q	11+	Clear	Bodied Oil	1.6	181.2	45.3	154.3	1.54	0.006	1.4852	0.9592
Linseed Mannitol.....	T	10	Clear	Normal	31.3	179.7	7.3	131.8	2.05	0.000	1.4862	0.9654
Linseed Sorbitol.....	P	9	Clear	Bodied Oil	18.2	185.0	8.0	146.8	1.63	0.018	1.4857	0.9606
Soybean Polyenta.....	P	10	Clear	Bodied Oil	1.2	177.0	37.4	129.8	0.89	0.013	1.4819	0.9553
Soybean Maleic Glyc.....	Q	8	Clear	Normal	6.3	212.6	37.5	120.1	0.14	0.000	1.4800	0.9592
Soybean Maleic Penta.....	S-	9	Clear	Normal	14.5	221.0	24.1	120.5	1.63	0.005	1.4806	0.9614
Z2 Bodied Linseed.....	Z2	6	Clear	Bodied Oil	6.6	194.0	4.8	123.8	1.54	0.000	1.4890	0.9664
Linseed Maleic Glyc.....	Z1+	7-	Clear	Acrid	7.9	211.8	23.4	138.0	1.05	0.014	1.4879	0.9763
Linseed Maleic Penta.....	Z2+	7	Clear	Acrid	11.8	212.8	24.8	155.5	0.98	0.000	1.4872	0.9731

TABLE II
Oil and Oil Film Tests

	Bodging Test		Drying Test		Sward Hard. 48 Hr.	Flash Point °F.	Resin Compatibility		Reactivity (zinc oxide)			
	Minutes at 307°C.		Color at Z2	Set to Touch Hr.			Dryness 48 Hr.	Urea Formaldehyde	Mod. Alkyd.	Orig. Visc. Sec.	% Increase	
	Z2	Gel.									24 Hr.	1 Week
G Bodied Linseed.....	138	253	9-	3 3/4	9+	3	565	I	C	166.0	-11.5	-22.9
Conjugated Linseed.....	105	199	8	2 3/4	9	3	530	I	C	154.0	-2.6	-26.5
Linseed Penta.....	72	120	13	2 1/2	9+	3	495	C	I	208.0	3.9	-24.5
Dehydrated Castor.....	65	111	8	1 3/4	9+	2	460	C	C	110.0	60.0	60.0
Fatty Tall Oil.....	165	420+	15	4	9-	3	I	I	112.0	28.3	54.8
G Bodied Soybean.....	312	480+	11+	8	8+	3	555	I	I	145.0	-9.0	-33.8
Conjugated Soybean.....	260	450	11-	7	8-	3	530	I	C	190.0	19.0	-29.5
Soybean Penta.....	120	195	14+	3 3/4	9+	4	540	C	C	207.0	-27.0	-45.4
Soybean Mannitol.....	147	275	18	4	8	1	435	C	C	34.0	0	0
Soybean Sorbitol.....	225	348	18+	3	9-	4	465	C	C	38.0	57.9	1150.0
Q Bodied Linseed.....	102	221	9	3 1/2	9	2	565	I	C	69.6	13.5	44.7
Linseed Polypenta.....	48	83	13	1 1/4	9	3	515	I	I	49.4	-1.8	-7.9
Linseed Mannitol.....	38	120	14	3	8	2	450	C	C	104.0	70.0	70.0
Linseed Sorbitol.....	45	120	18-	2 1/2	9	2	470	I	C	153.0	Livered	Livered
Soybean Polypenta.....	73	123	13	2 1/4	9	3	530	I	I	60.0	-10.8	-14.0
Soybean Maleic Glyc.....	150	257	11+	5	9	4	480	C	C	87.0	77.0	84.0
Soybean Maleic Penta.....	76	130	13	6	10-	5	585	C	C	55.5	-3.1	-9.2
Z2 Bodied Linseed.....	126	3 1/2	9	2	530	I	I	110.0	0	96.0
Linseed Maleic Glyc.....	58	2 1/2	10-	2	520	I	C	52.0	104.0	144.0
Linseed Maleic Penta.....	16	2	10-	4	470	I	C	109.0	292.0	787.0

Soybean Polypenta.—Technical poly-pentaerythritol esterified with nearly equivalent amounts of soybean oil fatty acids.

Soy. Maleic Glyc.—A soybean oil adduct made with 5% maleic anhydride and esterified with a near equivalent amount of glycerine.

Soy. Maleic Penta.—A soybean oil adduct made with 5% maleic anhydride and esterified with a near equivalent amount of technical pentaerythritol.

Z2 Bodied Linseed—An alkali refined, bleached, and refrigerated linseed oil heat bodied to Z2 viscosity.

Lin. Maleic Glyc.—A linseed oil adduct made with 5% maleic anhydride and esterified with a near equivalent of glycerine.

Lin. Maleic Penta.—A linseed oil adduct made with 5% maleic anhydride and esterified with a near equivalent of technical pentaerythritol.

The bodied linseed, bodied soybean, and dehydrated castor oils are included for the purpose of providing comparison controls for the synthetic and modified oils. Only one soybean oil control was used because in most cases the synthetic oils based on soybean fatty acids have performance characteristics which approximate or surpass those of linseed oil.

The analytical constants of the various oils are listed in Table I. Upon examination it will be seen

that they merely serve to characterize the oils and do not predict performance characteristics. The first four tests—viscosity, color, appearance, and odor—are purely physical in nature and describe the outward appearance of the oils. The term "normal" as used to describe the odor of an oil refers to a faint, bland, fatty odor such as that of a good quality alkali refined linseed oil. With synthetic esters such as are being discussed in this paper, the acid value and acetyl value show the degree of completion attained in esterification. It will be noticed that the mannitol and sorbitol esters have an excess of acid while the pentaerythritol and poly-pentaerythritol esters and the maleic treated oils contain some unreacted hydroxyl groups. The maleic treated oils also display a characteristically high saponification value (over 210). Because a certain amount of heat bodying always occurs during esterification, the iodine value of a synthetic ester is not such a reliable indication of quality as it is with the natural glycerides. The unsaponifiable and ash content of Fatty Tall Oil is much higher than that of the other oils as might be expected. The last two constants, refractive index and specific grav-

TABLE III
Oil and Oil Film Tests

	Film Solubility				Cold Water Resistance		Hot Water Resistance		Alkali Resist.			
	Water		Hexane	Acetone	Alcohol Benzene		Time to Whiten Hr.	Time to Fail Hr.		Time to Whiten Min.	Failure	Time to Fail Min.
	Per Cent	Acid Value			Per Cent	Acid Value						
G Bodied Linseed.....	9.4	193.9	20.2	44.7	48.8	104.2	140	140	35	Dull	32	
Conjugated Linseed.....	11.2	212.1	21.8	64.6	71.2	94.0	24	9	Broken blisters	11	
Linseed Penta.....	10.8	164.0	18.1	34.4	37.2	116.4	24	8	Dull, sl. white	8	
Dehydrated Castor.....	16.7	178.8	22.4	80.4	86.5	132.0	16	5	Film removed	9	
Fatty Tall Oil.....	14.9	128.0	24.4	75.0	88.4	100.4	18	42	1	Broken blisters	90	
G Bodied Soybean.....	15.9	118.4	31.1	97.7	98.2	89.7	16	5	Soft, dull, sl. white	56	
Conjugated Soybean.....	18.4	144.2	34.2	98.2	98.7	93.7	16	16	45	White, dull	100	
Soybean Penta.....	10.2	182.0	23.8	56.2	60.8	111.9	16	5	Soft, dull, sl. white	34	
Soybean Mannitol.....	16.7	207.8	28.0	94.0	95.5	120.5	16	16	7	Film removed	9	
Soybean Sorbitol.....	12.8	211.8	30.1	91.0	92.3	103.3	16	7	Film removed	14	
Q Bodied Linseed.....	9.7	178.0	17.0	42.4	46.9	101.8	141	142	36	Dull	34	
Linseed Polypenta.....	12.8	149.7	9.9	17.8	18.0	141.2	190	190	Broken blisters	2	
Linseed Mannitol.....	12.3	195.0	20.9	49.6	56.4	113.9	24	40	10	Soft, white, dull	5	
Linseed Sorbitol.....	11.4	171.5	17.4	38.0	41.1	120.4	16	7	White, dull, v. soft	11	
Soybean Polypenta.....	10.2	120.8	10.5	20.4	22.4	169.8	190	190	Broken blisters	2	
Soybean Maleic Glyc.....	9.2	187.5	21.6	40.5	43.4	120.8	16	7	Pass	14	
Soybean Maleic Penta.....	6.6	357.5	20.5	35.0	37.5	152.1	152.1	Pass	17	
Z2 Bodied Linseed.....	6.3	194.6	18.0	40.0	46.5	96.6	144	144	40	Dull	35	
Linseed Maleic Glyc.....	7.7	224.3	16.6	34.2	39.1	101.5	40	40	Dull	9	
Linseed Maleic Penta.....	10.3	187.0	18.2	36.7	40.0	112.9	40	40	Pass	5	

TABLE IV
 Bakelite BR 254 Varnishes

	Cooking Time Min.	Cooking Loss %	Visc.	Acid Value	Kauri Red. Pass %	Drying Test			Sward Hardness	Cold Water Resist. Time to Fail Hr.	Hot Water Resistance Failure	Alkali Resist. Time to Fail Hr.
						Set to Touch Min.	Dust Free Min.	Dryness 48 Hr.				
G Bodied Linseed.....	150	10.0	E	17.9	130	30	50	9+	7	> 168	Dull	46
Conjugated Linseed.....	120	10.0	D	15.3	220	21	30	9	5	> 168	White, dull	19
Linseed Penta.....	135	6.1	D	19.4	150	25	58	9	4	> 192	Pass	114
Dehydrated Castor.....	86	9.2	E	22.7	220	20	28	10	7	> 168	Pass	66
Fatty Tall Oil.....
G Bodied Soybean.....	285	13.1	D	13.2	140	30	50	9	4	> 192	Pass	40
Conjugated Soybean.....	285	12.6	D	13.6	170	25	35	9	2	> 190	Soft, dull	4
Soybean Penta.....	150	6.1	D-	15.1	230	20	65	8+	2	> 192	Pass	66
Soybean Mannitol.....	195	15.0	D-	22.0	140	23	60	9+	4	> 190	Pass	19
Soybean Sorbitol.....	190	15.0	D	17.0	140	28	45	9	4	> 192	Pass	27
Q Bodied Linseed.....	155	8.2	D+	18.7	160	41	61	10-	4	> 168	Pass	4
Linseed Polypenta.....	31	3.7	D+	21.8	180	28	330	10-	7	> 192	Broken blisters	288
Linseed Mannitol.....	120	12.6	E+	28.2	120	43	53	10	11	> 190	Pass	199
Linseed Sorbitol.....	95	9.7	D+	24.3	90	20	40	10-	14	> 192	Soft, white, dull	40
Soybean Polypenta.....	60	4.4	D	20.0	190	30	340	9	3	> 192	Broken blisters	288
Soybean Maleic Glyc.....	165	7.6	D	20.6	150	25	60	10-	6	> 196	Soft, dull	4
Soybean Maleic Penta.....	47	8.0	D-	28.8	90	75	328	10-	10	> 192	Pass	4
Z2 Bodied Linseed.....	60	4.2	D	27.6	130	253	450	7+	6	> 168	Broken blisters	27
Linseed Maleic Glyc.....
Linseed Maleic Penta.....

ity, are of a purely physical nature and vary more with the viscosity of the oil than with its composition.

Table II lists some of the performance characteristics of the oils and oil films. An examination of the results of the bodying test shows that the polypentaerythritol esters are much faster bodying than the other synthetic esters. The pentaerythritol esters are next and the mannitol and sorbitol esters are the slowest bodying. Soybean Sorbitol is much slower bodying than Soybean Mannitol, but no such difference occurs with the corresponding linseed esters. The sorbitol and mannitol esters also produce much darker bodied oils than the other synthetic oils. The maleic treated oils which have been esterified with pentaerythritol are much faster bodying than those esterified with glycerine. Most of the oils have about the same flash and fire points, except for the sorbitol and mannitol esters which are much lower. The drying times of the synthetic esters follow the same order as the bodying rates and all are faster than the corresponding natural glycerides. The fastest drying time is shown by the polypentaerythritol esters, fol-

lowed by pentaerythritol, sorbitol, and mannitol in that order. None of the synthetic oils dried a great deal harder than the natural glycerides and the mannitol and sorbitol esters were somewhat softer. Resin compatibility is not a particularly important test from a performance standpoint. It is intended only to discover any unusual compatibility characteristics of the oils which might be made the basis of some novel practical applications.

Oil and oil film properties are continued in Table III. The film solubility test is designed to give an insight into the film structure. The amount of low molecular weight, water soluble materials produced is approximately the same for all oils. Variations between the oils become apparent in the amounts of slightly higher molecular weight, hexane soluble products and become most apparent in the acetone and alcohol-benzene soluble fractions. The last two solvents appear to remove not only the decomposition products but also uncombined and loosely bound oil molecules. The amount of material which is soluble in acetone or alcohol-benzene varies directly with the

 TABLE V
 Amberol 801 Varnishes

	Cooking Time Min.	Cooking Loss %	Visc.	Acid Value	Kauri Red. Pass %	Drying Test			Sward Hard. 48 Hr.	Cold Water Resist. Time to Fail Hr.	Hot Water Resist. Failure	Alkali Resist. Time to Fail Min.
						Set to Touch Min.	Dust Free Min.	Dryness 48 Hr.				
G Bodied Linseed.....	58	1.1	D	11.9	90	133	140	10	43	> 168	Pass	40
Conjugated Linseed.....	104	1.1	E+	11.0	100	60	74	10	6	114	Dull	20
Linseed Penta.....	68	3.2	E+	9.2	100	58	84	10	21	> 120	Pass	32
Dehydrated Castor.....	43	1.8	F+	11.4	130	56	76	10	6	190	Soft, dull	15
Fatty Tall Oil.....
G Bodied Soybean.....	128	2.9	F	11.7	190	61	87	9	6	> 120	Pass	30
Conjugated Soybean.....	95	1.8	E	10.0	140	63	81	9	2	114	Soft, dull	36
Soybean Penta.....	67	2.1	F	8.8	90	75	105	9+	12	> 112	White, dull	80
Soybean Mannitol.....	26	1.3	D	12.3	70	75	105	9+	17	112	White, dull	34
Soybean Sorbitol.....	80	1.6	D	14.4	70	75	105	10-	17	112	Dull	27
Q Bodied Linseed.....	58	1.1	D	11.9	90	133	140	10	43	> 168	Pass	40
Linseed Polypenta.....************
Linseed Mannitol.....	75	4.2	D+	16.2	80	62	77	10	17	114	Very dull	37
Linseed Sorbitol.....	69	4.3	F+	13.4	40	70	100	10	18	> 112	White, dull	32
Soybean Polypenta.....************
Soybean Maleic Glyc.....	36	3.2	D+	11.7	120	58	89	10	8	> 120	Pass	15
Soybean Maleic Penta.....	0	1.3	D-	15.3	50	140	246	10-	10	> 192	Pass	50
Z2 Bodied Linseed.....	58	1.1	D	11.9	90	133	140	10	43	> 168	Pass	40
Linseed Maleic Glyc.....	25	1.3	F	13.6	100	57	77	10	9	190	Soft, dull	20
Linseed Maleic Penta.....	28	3.9	E	15.5	80	65	85	10	12	190	Soft, dull	14

* Incompatible.

TABLE VI
Limed Rosin Varnishes

	Cooking Time Min.	Cooking Loss %	* Visc.	Acid Value	Kauri Red. Pass %	Drying Test			Sward Hard. 48 Hr.	Cold Water Resist. Time to Fail Hr.	Hot Water Resistance Failure	Alkali Resist. Time to Fail Min.
						Set to Touch Min.	Dust Free Min.	Dryness 48 Hr.				
G Bodied Linseed.....	105	6.1	D+	18.3	90	120	180	9+	36	48	White, dull	25
Conjugated Linseed.....	75	6.9	E	14.5	100	53	100	10-	16	43	Broken blisters	6
Linseed Penta.....	30	2.9	D+	16.4	90	45	115	10-	14	112	Blistered	62
Dehydrated Castor.....	75	6.8	F	18.4	100	58	120	10-	11	43	White, dull	12
Fatty Tall Oil.....
G Bodied Soybean.....	195	8.4	E	16.8	200	56	216	8+	3	43	White, dull	7
Conjugated Soybean.....	75	6.1	D	15.3	120	47	100	9	6	43	Film removed	4
Soybean Penta.....	115	5.0	E	13.7	130	75	105	10-	10	39	White, dull	41
Soybean Mannitol.....	115	11.8	D+	17.1	90	70	100	10-	16	64	White, dull	10
Soybean Sorbitol.....	125	13.7	E	18.7	110	75	105	10-	14	24	White, dull	13
Q Bodied Linseed.....	105	6.1	D+	18.3	90	120	180	9+	36	48	White, dull	25
Linseed Polyenta.....	35	6.4	D+	16.6	90	46	338	10	11	19	Broken blisters	5
Linseed Mannitol.....	100	10.0	F+	27.2	80	52	72	10	18	120	Dull	1
Linseed Sorbitol.....	90	11.8	F	22.8	60	85	115	10-	21	64	White, dull	22
Soybean Polyenta.....	48	5.3	E+	15.4	100	87	348	9	7	19	Broken blisters	5
Soybean Maleic Glyc.....************
Soybean Maleic Penta.....*	11.6	D-	12.0	70	140	246	10	13	> 192	Broken blisters	157
Z2 Bodied Linseed.....	105	6.1	D+	18.3	90	120	180	9+	36	48	White, dull	25
Linseed Maleic Glyc.....************
Linseed Maleic Penta.....	50	8.2	D	11.3	80	42	62	10	18	47	Broken blisters	20

* Incompatible.

functionality of the alcohol and the unsaturation present in the fatty acids and, with alcohol-benzene, ranges from a low of 18.0% for Linseed Polyentaerythritol to a high of 98.2% for G Bodied Soybean. It may be remarked that the hexahydric alcohols, mannitol and sorbitol, exhibit greater solubility than the tetrahydric pentaerythritol, but it must be remembered that mannitol and sorbitol both lose water with the formation of inner ethers at esterification temperatures, so that their actual functionality is usually less than four. In most cases the reactivity of the oils with zinc oxide was quite satisfactory, only the sorbitol esters and Linseed Maleic Pentaerythritol showed excessively large increases in viscosity. The polyentaerythritol esters showed the greatest resistance to cold and hot water, followed by pentaerythritol, mannitol, and sorbitol in that order. The soybean maleic oils had slightly better cold and hot water resistance than the linseed maleic oils. No

particular trend can be noticed in the results of the alkali resistance tests.

Table IV shows the characteristics of varnishes made from the various oils and a pure phenolic resin, Bakelite BR 254. The polyentaerythritol esters gave the fastest cooking times, followed by the pentaerythritol, sorbitol, and mannitol esters in that order. Soybean Maleic Pentaerythritol was faster cooking than Soybean Polyentaerythritol while Soybean Maleic Glycerine required about the same time as Soybean Pentaerythritol. No varnishes were made with the linseed maleic oils because these oils both had Z2 viscosities while the resin requires a lower viscosity oil for good results. Most of the oils had excellent Kauri reduction values, only two of them, Linseed Sorbitol and Soybean Maleic Pentaerythritol, failing as low as 100%. The various esters do not produce any great differences in the drying times of the varnishes made from them, but the varnishes made with sorbitol

TABLE VII
Paint Tests

	Application		Consistency		Drying Odor	Dry to Touch			Hardness			
	S. P.	M. P.	1 Day	7 Days		Hours at 77°F.	Hours at 40°F.	Min. at 230°F.	1 Day		3 days	
			M. P.	M. P.					S. P.	M. P.	S. P.	M. P.
G Bodied Linseed.....	Good	Good	183	180	Mod.	6½	42	6½	V. soft	Soft	Hard	Hard
Conjugated Linseed.....	V. good	V. good	195	198	Mod.	6¼	50	14	Soft	Firm	Soft	Hard
Linseed Penta.....	Good	Good	219	220	Mod.	4½	26	7½	Soft	Firm	Firm	V. hard
Dehydrated Castor.....	Good	Good	260	260	Mod.	3½	26	10	Firm	Hard	Hard	V. hard
Fatty Tall Oil.....	Good	Good	204	208	Mod.	8+*	25	Soft	Soft	Firm	Hard
G Bodied Soybean.....	Good	Good	186	184	Mod.	8+*	19	Soft	Soft	Soft	Firm
Conjugated Soybean.....	V. good	V. good	197	207	Mod.	8+*	15	V. soft	Soft	Soft	Firm
Soybean Penta.....	Good	Good	205	212	Mod.	4½	42	10	V. soft	Firm	Firm	Hard
Soybean Mannitol.....	Bad	Poor	700+	700+	Mod.	7½	42	21	V. soft	Firm	Soft	Hard
Soybean Sorbitol.....	Fair	Fair	700+	700+	Mod.	7¾	42	17	Soft	Firm	Hard	Hard
Q Bodied Linseed.....	Fair	Good	381	400	Slight	6	26	6	Soft	Soft	Hard	Hard
Linseed Polyenta.....	Good	Fair	230	220	Strong	3½	18	6	Firm	Firm	Hard	Hard
Linseed Mannitol.....	Fair	Poor	700+	700+	Mod.	4¾	26	12	Soft	Soft	Hard	Firm
Linseed Sorbitol.....	Poor	Fair	408	401	Slight	5¾	26	6½	V. soft	Soft	Firm	Hard
Soybean Polyenta.....	Good	Fair	230	225	Strong	3½	18	6	Hard	Hard	Hard	V. hard
Soybean Maleic Glyc.....	Good	Fair	310	339	Mod.	5¼*	6½	Soft	Firm	Firm	Hard
Soybean Maleic Penta.....	Good	Fair	275	302	Mod.	5½	50	7	Soft	Firm	Firm	Hard
Z2 Bodied Linseed.....	V. good	V. good	205	215	Slight	4½	42	7	Soft	Hard	Firm	Hard
Linseed Maleic Glyc.....	Good	V. good	288	309	Mod.	3¾	42	4	Firm	Hard	Hard	V. hard
Linseed Maleic Penta.....	Good	V. good	248	268	Mod.	2¾	26	3½	Firm	Hard	Hard	Hard

* Didn't dry.

esters produced the hardest films, followed by mannitol, polypentaerythritol, and pentaerythritol. Both of the soybean maleic oils produced harder films than the soybean esters. The cold water and hot water resistances of the varnishes were generally good. Both of the polypentaerythritol ester varnishes failed in hot water resistance because of broken blisters but showed no sign of whitening. The alkali resistance of the polypentaerythritol ester varnishes was outstanding, followed by pentaerythritol, sorbitol, and mannitol in decreasing order. The soybean maleic varnishes had less alkali resistance than any of the varnishes made with soybean esters.

Table V shows the characteristics of varnishes made from various oils and a rosin-maleic resin, Amberol 801. All the oils were bodied to Z2 viscosity before cooking these varnishes. For this reason it was only necessary to prepare one varnish for the three bodied linseed oils. The maleic treated oils were the fastest cooking group. Most of the esters required approximately the same cooking times, the exceptions being the polypentaerythritols, which were incompatible, and Soybean Mannitol, which had a much shorter cooking time than the other esters. All of the varnishes made with synthetic oils dried rapidly and produced hard films. The pentaerythritol esters produced about the best results, but were not greatly superior to the other esters. Soybean Maleic Pentaerythritol produced a varnish that remained slightly tacky but was as hard as those made with the other soybean base synthetic oils. The soybean maleic oils and Linseed Pentaerythritol produced varnishes having excellent cold water and hot water resistances. The Soybean Pentaerythritol and Soybean Maleic Pentaerythritol varnishes had excellent alkali resistance. The other synthetic oils produced varnishes whose alkali resistances were not quite as good as that of the Z2 Bodied Linseed varnish.

Table VI shows the characteristics of varnishes made from the various oils and 4% Lined Rosin. All the oils were bodied to Z2 viscosity before cooking these varnishes and, as with the Amberol 801 varnishes, only one varnish was necessary for the various bodied linseed oils. Linseed Maleic Glycerine and Soybean Maleic Glycerine were incompatible

with Lined Rosin. Linseed Poly-pentaerythritol and Linseed Pentaerythritol are much faster cooking than the other linseed base synthetic oils, and Soybean Poly-pentaerythritol is much faster than the other soybean base oils. The varnishes made with the synthetic oils dried faster than those made with the natural oils and those based on soybean oil also dried harder than the varnish made with G Bodied Soybean. The polypentaerythritol esters produced varnishes having less cold water but more hot water resistance than those made with the natural oils, while the other synthetic oils tended to produce greater cold water and about equal hot water resistances. The greatest alkali resistances were produced by the varnishes made with the pentaerythritol esters and Soybean Maleic Pentaerythritol.

Table VII lists some of the properties of paints made with the oils. Two paints were made with each oil, one a single pigment type using zinc sulfide as the sole pigment and the other a mixed pigment paint. The application test, which includes appearance, brushing, leveling, and gloss records a master painter's opinion of the handling qualities of the paints. Most of the paints had fair to good consistencies but those of paints made with the sorbitol and mannitol esters are excessively high. The drying times of the paints corresponded to the drying times of the oils from which they were made. The paints made with polypentaerythritol esters were the fastest drying, followed by those made with pentaerythritol, sorbitol, and mannitol esters in that order. All of the paints made with synthetic oils dried faster than those made with the corresponding natural oils.

Table VIII lists some further characteristics of the paints. The results of the after yellowing, water permeability, and flexibility tests do not show any pronounced trends. In general, the films of paints made with the synthetic oils had less elongation than those of the paints made with natural oils. The two maleic glycerine oils seem to have a tendency to promote brittleness. All of the linseed base synthetic oils produced greater tensile strength than bodied linseed oil of the same viscosities. The two linseed maleic oils produced greater tensile strength in one formulation and less in another. In the mixed pig-

TABLE VIII
Paint Tests

	Water Permeability M. P.	Flexibility M. P.	Elongation Stripped Film		Adhesion S. P.	Tensile Strength of Stripped Film		Taber Abrasion M. P.	Shear Hardness M. P.	60° Gloss M. P.	Daylight Reflect. M. P.	After Yellowing M. P.
			S. P.	M. P.		S. P.	M. P.					
G Bodied Linseed.....	0.57	28+	34.0	8.0	Fair	5.0	20.0	339	500	35.5	81.0	Considerable
Conjugated Linseed.....	0.63	28+	34.0	6.0	V. poor	8.0	27.0	344	405	45.5	80.3	Slight
Linseed Penta.....	0.59	28+	18.0	5.0	Poor	20.0	37.0	200	583	26.0	81.0	V. Slight
Dehydrated Castor.....	0.64	28+	25.5	9.0	Poor	14.7	38.7	264	428	36.2	81.0	V. Slight
Fatty Tall Oil.....	0.54	28+	25.5	3.1	Fair	11.0	16.3	310	637	29.3	80.0	Slight
G Bodied Soybean.....	0.52	28+	84.0	8.0	Fair	0.0	9.0	724	234	30.5	81.7	Slight
Conjugated Soybean.....	0.76	28+	80.0	7.0	Fair	0.0	0.0	930	182	34.0	80.6	V. Slight
Soybean Penta.....	0.94	16.0	12.5	5.0	Fair	11.0	30.0	399	373	14.5	83.0	Slight
Soybean Mannitol.....	0.55	18.0	60.0	3.0	Fair	0.0	10.0	395	278	18.3	80.8	None
Soybean Sorbitol.....	0.53	19.0	39.0	3.0	Good	8.0	18.0	331	396	17.7	82.9	V. Slight
Q Bodied Linseed.....	0.57	28+	24.0	7.0	Good	13.0	32.0	254	145	69.0	80.2	V. Slight
Linseed Poly-penta.....	0.54	20.0	6.0	1.5	Good	33.3	58.7	215	625	27.0	82.3	Considerable
Linseed Mannitol.....	0.50	28+	36.0	4.0	Fair	12.0	30.0	216	595	27.0	81.0	V. slight
Linseed Sorbitol.....	0.49	14.0	21.0	4.0	Good	18.0	48.0	199	524	32.5	78.8	V. slight
Soybean Poly-penta.....	0.56	25.0	10.5	1.5*	Good	25.5	36.9	227	645	27.5	83.3	Slight
Soybean Maleic Glyc.....	0.63	14.0	17.0	3.0*	V. poor	0.0	25.0	370	362	46.0	78.5	None
Soybean Maleic Penta.....	0.67	28+	12.5	6.0	Poor	30.8	40.9	320	1000	30.4	81.2	V. Slight
Z2 Bodied Linseed.....	0.45	16.0	32.0	2.0	Good	19.0	29.0	236	536	13.0	81.5	None
Linseed Maleic Glyc.....	0.52	13.0	18.0	2.0	Poor	14.0	38.0	218	597	15.7	82.0	Slight
Linseed Maleic Penta.....	0.53	12.0	16.0	0.0*	Poor	28.0	22.0	250	730	12.5	82.1	Slight

* Brittle.

ment paints, the soybean synthetics produced greater tensile strength than G Bodied Soybean, while in the single pigment paints there was in several cases no difference. The paints made with Soybean Maleic Pentaerythritol and the two polypentaerythritol esters had tensile strengths which were much superior to those of the paints made with natural oils. Conjugated Soybean did not have any tensile strength in either formulation. Abrasion resistance was not determined on the single pigment paints because they were too soft. Abrasion resistance of paints made with the various oils showed only that most of the synthetics were better than the corresponding natural oils. Conjugated Soybean, however, had less abrasion resistance than G Bodied Soybean. In most cases the shear and Sward hardnesses of the paints made with synthetic oils were greater than those of the paints made with the natural oils. Soybean Maleic Pen-

taerythritol and the two polypentaerythritol esters produced particularly hard films. Gloss measurements on the paint films show no particular trends.

Daylight reflectance is a measure of the whiteness of the paints and also shows no particular trends.

In a following paper it is intended to present a method whereby the experimental data gathered under this evaluation scheme may be condensed and correlated for convenience in reference and comparison.

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REFERENCE

1. Bolley, Don S., and Gallagher, E. C., *J. Am. Oil Chem. Soc.* **24**, 146 (1947).

☆ ☆ ☆ **ABSTRACTS** ☆ ☆ ☆

Oils and Fats

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CHEMICAL TREATMENTS OF SEEDS TO PREVENT HEATING AND DETERIORATION DURING STORAGE. A. M. Altschul (So. Reg. Res. Lab., New Orleans, La.). *Cotton Gin & Oil Mill Press*, Jan. 8, 1949. Forty-eight chemicals which inhibit heating and deterioration of seeds during storage are recorded. Mixtures of 2 chemicals were more efficient than either one used alone.

SOLVENT EXTRACTION OF SOYBEAN OIL BY MIXTURES OF TRICHLOROETHYLENE AND ETHYLENE ALCOHOL. S. G. Measmer, O. R. Sweeney, and L. K. Arnold. *Proc. Iowa Acad. Sci.* **54**, 189-97(1947). A solvent consisting of denatured alcohol mixed with trichloroethylene to give a specific gravity of 0.910 can be used successfully in a continuous extraction plant to remove oil from soybeans. The extraction is carried out at 70° and the miscella cooled to cause it to separate into two phases. The lower phase can be separated and stripped to remove the solvent while the upper phase can be returned to the system without evaporation to extract more oil. The moisture content in the solvent will not build up if beans having less than 6% moisture are used and the drying is done with 10-lb. per square inch steam pressure on the drier. (*Chem. Abs.* **43**, 1999.)

PYROLYSIS OF PALM OIL WASTE. R. Francois. *Oleagineux* **3**, 602-6(1948). The possibility of manufacture of coke, tar, gas, fertilizer, oil, etc. from palm oil cake and pulp is discussed from laboratory tests on the subject.

ALCOHOLS BY SODIUM REDUCTION. A STAFF-INDUSTRY COLLABORATIVE REPORT. M. L. Kastens and H. Peddicord. *Ind. & Eng. Chem.* **41**, 438-46(1949).

UNSATURATED LONG-CHAIN ALIPHATIC ESTERS. CATALYTIC OXIDATION. T. M. Patrick and W. S. Emerson (Monsanto Chem. Co., Dayton, Ohio). *Ind. Eng. Chem.* **41**, 636-41(1949). The liquid-phase air oxidation of the Me esters of tall oil fatty acids is described. In order to develop a technic for the method, oxidation of Et oleate and of oleic acid was studied in some detail. Oxidative scission was expected to take place

at or near the double bonds to yield mono- and dicarboxylic acids. Large amounts of high molecular weight by-products, in addition to the expected scission products, were formed in these oxidations. The identified products comprised most of the members of the homologous series from caproic to palmitic acid and from suberic to undecanedioic acid, inclusive.

THE NITROGENOUS CONSTITUENTS OF THE TISSUE LIPIDES. I. THE EXTRACTION, PURIFICATION, AND HYDROLYSIS OF TISSUE LIPIDES. J. M. McKibbin and W. E. Taylor (Syracuse Univ. College Med.). *J. Biol. Chem.* **178**, 17-27(1949). A technic for the determination of total lipide nitrogen is described. The amounts of nitrogen, phosphorus, and choline present in lipide extracts of 9 tissues from several dogs are presented.

THE NITROGENOUS CONSTITUENTS OF THE TISSUE LIPIDES. II. THE DETERMINATION OF SPHINGOSINE IN TISSUE LIPIDE EXTRACTS. J. M. McKibbin and W. E. Taylor (Syracuse Univ.). *J. Biol. Chem.* **178**, 29-35(1949). A method is described for an approximate determination of sphingosine in whole lipide extracts which have been freed of non-lipide impurities. The method is based on a chloroform extraction of the aqueous lipide hydrolysate which is both specific and relatively quantitative for sphingosine. Sphingosine is then determined on the chloroform extract as nitrogen.

TESTING EMULSIONS OF VEGETABLE OILS BY MEANS OF ULTRASONIC WAVES. A. Audouin and G. Levavasseur. *Oleagineux* **4**, 95-100(1949). The authors treat the influence of the frequency of sound waves in the production and rupture of water-in-oil and oil-in-water emulsions.

OXIDATION OF ESTERS OF LINOLEIC ACID BY OXYGEN. W. O. Lundberg (Hormel Inst., Austin, Minnesota). *Oleagineux* **4**, 86-93(1949).

AN OXIDATION-REDUCTION CYCLE IN EMULSION POLYMERIZATION SYSTEMS. F. T. Wall and T. J. Swoboda (Univ. Illinois). *J. Am. Chem. Soc.* **71**, 919-24(1949).